Arrhenius plot: $k \ (\text{cm}^3/\text{mol}/\text{s}) \ vs \ 1/T \ (\text{K}^{-1})$

- $N+NO=N_2+O \ (p164)$
- $N+NO=N_2+O \ (p171)$
- $O+H_2=OH+H \ (p112)$