

## 2.2 Interfacial Tension and Wettability

Interfacial tension due to molecular attraction - different in two fluids

Interfacial tension,  $\sigma_{ik}$ , between two fluids  $i$  and  $k$ .

$\sigma_{ik}$  is temperature dependent  $\therefore$  Capillary pressure =  $f(T)$ .

### Dupré's formula

$$W_{ik} = \sigma_i + \sigma_k - \sigma_{ik} \Rightarrow \text{work to separate into substances } i \text{ and } k \text{ with vapor interface}$$

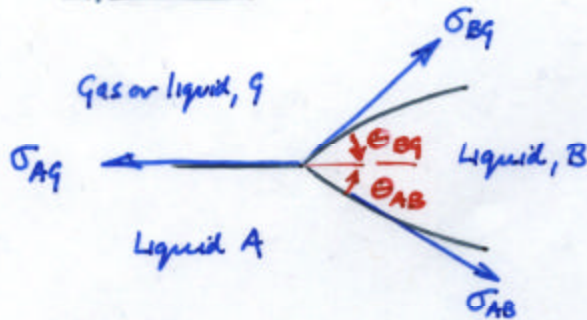
$\sigma_i$  = surface tension

$\sigma_{ik}$  = interfacial tension

$$\frac{i}{k} \Rightarrow \frac{\text{vapor } i}{\text{vapor } k}$$

$\sigma_i$  = surface tension of fluid with its own vapor.

### Equilibrium



$$\text{Equilibrium: } \sigma_{AG} = \sigma_{AB} \cos \theta_{AB} + \sigma_{BG} \cos \theta_{BG}$$

Only satisfied if  $\sigma_{AG} < (\sigma_{AB} + \sigma_{BG})$

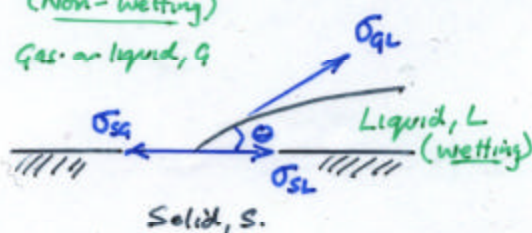
$\therefore$  lens of B formed.

If  $\sigma_{AG} > (\sigma_{AB} + \sigma_{BG})$

then B spreads between A and G.

(Non-wetting)

Gas or liquid, G



By convention,  $\theta$  measured in denser fluid

$$\text{Equilibrium: } \sigma_{GL} \cos \theta = \sigma_{SG} - \sigma_{SL}$$

$$\cos \theta = (\sigma_{SG} - \sigma_{SL}) / \sigma_{GL}$$

$\therefore$  May determine interface angle for known  $\sigma$  (contact angle,  $\theta$ ).

- influenced by surface roughness.

If  $(\sigma_{sq} - \sigma_{sl})/\sigma_{ql} > 1 \Rightarrow$  No equilibrium  
Liquid flows over solid  $\rightarrow$  wettability

### Wettability

Defined by  $\theta$

$\theta < 90^\circ$  eg Fluid, L wets the solid; wetting fluid

$\theta > 90^\circ$  eg Fluid G, is non wetting fluid

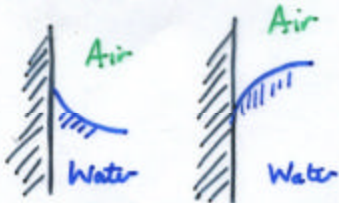
$\theta = 90^\circ$  Zero adhesion tension since equal affinity.

Wettability is controlled by: a) Chemical composition of fluids  
eg. adding dopant

b) Sequence of wetting

eg. Advancing or receding.

$\therefore$  hysteresis - wetting or draining



Water wets  
solid

Water non-wets  
solid

Most geologic systems are water wet.

petroleum/water

air/water

NAPL/water

## IMPORTANT ASPECTS OF WETTING

### Water-Wet

- a) Pendular rings @ grain contacts
- no continuous water-phase
  - except mono-molecular coating of  $H_2O$
- b)  $\uparrow$  wetting phase saturation  $\rightarrow$
- continuous water (wetting) phase
  - equilibrium saturation = when phase is continuous  
 $\uparrow$   
(wetting)

Implication: wetting phase may be removed.

- c) Above critical saturation (equilibrium) funicular saturation
- Now non-wet phase non-continuous.
  - Can only remove non-wetting phase if large pressure gradient to squeeze through pore throat.

### Funicular saturation critical in remediation!

Remediation @ residual saturations:

Important since difficult to remove last few % of saturation.

Grains



Pools



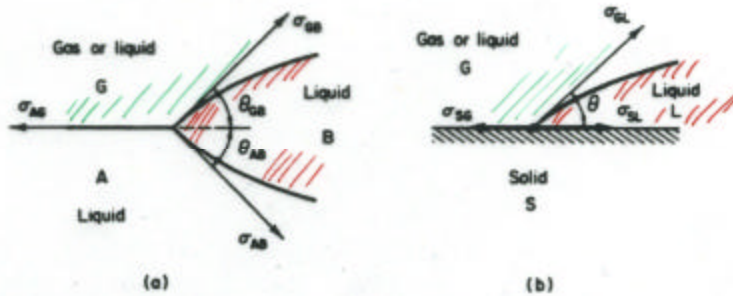


FIG. 9.2.1. Interfacial tensions.

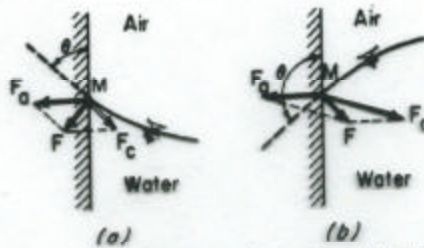


FIG. 9.2.2. Contact angle between a water-air interface and a solid. (a) Water wetting the solid. (b) Water nonwetting the solid.

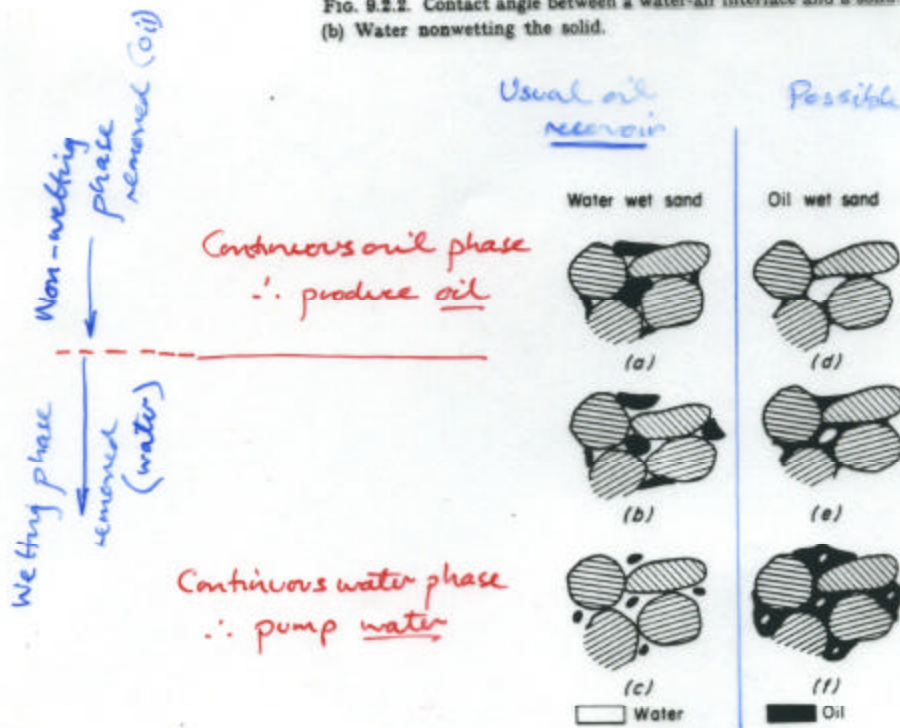


FIG. 9.2.3. Possible fluid saturation states.