









- Evaluate the effective water and TCE saturations at the locations of each of the piezometers. Explain your rationale for choice of drainage or imbibition curves.

4. What is the permeability of the silt to TCE? State your assumptions.

5. What is the volumetric flow rate in the vertical direction per plan area of flow?

### Question 3

A two-component DNAPL cocktail has been spilled through the vadose zone in a sand aquifer and has reached and penetrated the saturated zone. Soil samples are taken from the saturated zone within (an arbitrary) part of the aquifer where a dissolved plume is presumed to have developed. The sample is centrifuged to remove the pore fluids, and the fluid assayed to determine aqueous concentrations,  $c_a$ . Components are desorbed from the solid grains to define the presumed equilibrium sorbed concentrations,  $c_s$ . The porosity of the sand aquifer is  $n=25\%$ , the bulk density is  $\rho_b=1200 \text{ kg/m}^3$ , and the mean volumetric moisture content in the vadose zone is  $\theta=5\%$ . The mean soil temperature is  $20^\circ\text{C}$ . Aqueous solubility of each of the components approximately triple with an increase in temperature from  $20^\circ$  to  $60^\circ\text{C}$ .

Component	$c_a$ <i>mg/l</i>	$c_s$ <i>mg/kg</i>	Mole fraction, $X_i$ %
Trichloroethane (TCA)	$0.13 \times 10^3$	325	60
Methyl Chloride	$0.6 \times 10^4$	600	40

Component	Aqueous Conc. $c_a$ ( $60^\circ\text{C}$ ) <i>mg/l</i>
Trichloroethane (TCA)	
Methyl Chloride	

1. Complete the missing entries in the table above.

2. Approximately 10,000 l of the cocktail is to be removed from the system. Evaluate the time taken to remove this material from the aquifer if the system is flushed with water at 20°C, at a rate of 40 m<sup>3</sup>/day. Assume that mole fraction approximates mass fraction, and that mean density of the NAPL is 1400 kg/m<sup>3</sup>.

3. What is the time taken to remove the material if warm water is used.